CLAIMS:

- 1. (Currently Amended) A lithium-metal-oxide electrode material eempositions and structures—having a single-phase layered crystallographic structure and the general formula $\text{Li}_x \text{Mn}_y \text{M}_{z-y} \text{O}_2$ where $0 \le x \le 0.20$, $\text{Li}_x \text{Mn}_y \text{M}_z$ O_2 where $0 < x \le 0.24$, 0 < y < 1, z = 0.46 to 0.86, and y + z = 0.94 to 0.99, manganese is in the 4+ oxidation state and M is one or more transition metal or other cations, but is not solely Ni or Cr.
- 2. (Currently Amended) A material according to claim 1, wherein M is chosen from all of the other first row transition metals: Ti, V. Cr, Fe, Co, Ni and Cu, and other cations with appropriate sized ionic radii: Al, Mg, Mo, W, Ta, Si, Sn, Zr, Be, Ca, Ga, and P, but is not solely Ni.
- 3. (Currently Amended) A material according to claim 1, wherein M is one or more transition metal or other cations chosen from the other first row transition metals: Ti, V. Cr, Fe, Co, Ni and Cu, and other metal cations such as selected from Al, Mg, Mo, W, Ta, Ga and Zr.
- 4. (Original) A material according to claim 1, wherein M is one or more transition metal or other metal cations chosen from the first row transition metals and Al.
- 5. (Previously Amended) The use of a material according to claim 1, as positive electrode in a non-aqueous lithium cell or battery, such as a lithium ion cell.
- 6. (Currently Amended) A process for making a material of formula $\text{Li}_x \text{Mn}_y \text{M}_{1-y} \text{O}_2$, $\text{Li}_x \text{Mn}_y \text{M}_z \text{O}_2$ wherein $\text{x} \leq 0.2$, $0 < \text{x} \leq 0.24$, 0 < y < 2, 1, z = 0.46 to 0.86, and y + z = 0.94 to 0.99. Mn is Mn+4 and M is one or more transition metal cations or other cations, but is not solely Ni or Cr comprising providing a starting material of formula $\text{Li}_{1+x} \text{Mn}_y \text{M}_{1-y} \text{O}_2$, wherein x is equal to or greater than 0, $\text{Li}_x \text{Mn}_y \text{M}_z \text{O}_2$, M is one or more transition metal or other cations, as a cathode in a lithium ion cell, and charging the cell to a high voltage in the

voltage in the range of 4.4 to 5 volts versus the potential of metallic lithium in the temperature range of room temperature to 55°C.

- 7. (Original) A process according to claim 6, wherein M is chosen from all of the other first row transition metals: Ti, V. Cr, Fe, Co, Ni and Cu, and other cations with appropriate sized ionic radii: Al, Mg, Mo, W, Ta, Si, Sn, Zr, Be, Ca, Ga, and P, but is not solely Ni.
- 8. (Currently Amended) A process according to claim 6, wherein M is one or more transition metal or other metal cations chosen from the other first row transition metals: Ti, V. Cr, Fe, Co, Ni and Cu, and other cations such as selected from Al, Mg, Mo, W, Ta, Ga and Zr.
- 9. (Original) A process according to claim 6, wherein M is one or more transition metal or other metal cations chosen from the first row transition metals and Al.
- 10. (Cancelled)
- 11. (New) A process according to claim 6, wherein the temperature is room temperature.
- 12. (New) A process according to claim 6, wherein the starting material is $Li_{1.2}$ Mn_{0.4} M_{0.4} O₂ wherein M is Ni_{0.4-x} Co_x wherein x = 0.1 to 0.4.
- 13. (New) A process according to claim 12, wherein M is $Ni_{0.2}$ $Co_{0.16}$ $Cu_{0.4}$ or $Ni_{0.2}$ $Co_{0.1}$ $Al_{0.1}$.
- 14. (New) A material according to claim 1, wherein M is Ni, Co or Ti, Ni, Co or Ni, Co, Cu or Ni, Co, Mg or Ni, Co, Al.
- 15. (New) A material according to claim 14, wherein x=0.05 to 0.20.

- 16. (New) A material according to claim 1, wherein x=0.06 to 0.24 and y= 0.12 to 0.49.
- 17. (New) A process according to claim 6, wherein y'=0.118 to 0.4.